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pH

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pH

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( )

$Zn^{+2}$

pH

pH

( )

pH

( )

( )

( )

+ +

( )  $Mn^{+2}$

( )  $Mn^{+4}$

pH ( )

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pH

$$\frac{x}{m} = K_f C^{\frac{1}{n}}$$

$$\log \frac{x}{m} = \log K_f + \frac{1}{n} \log C$$

$$\frac{x}{m}$$

( )

:C

$$(\quad)$$

$$: k_f - n$$

$$\log C - \log \frac{x}{m}$$

$$\log \frac{1}{n} - k_f$$

)

$$\frac{1}{n} - K_f$$

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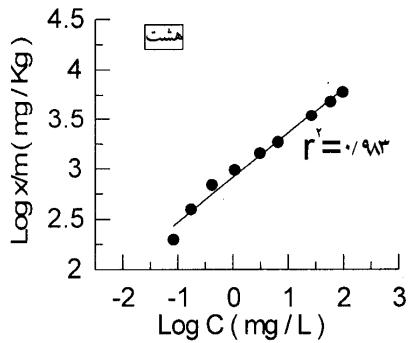
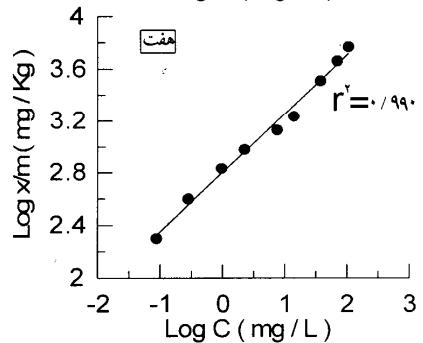
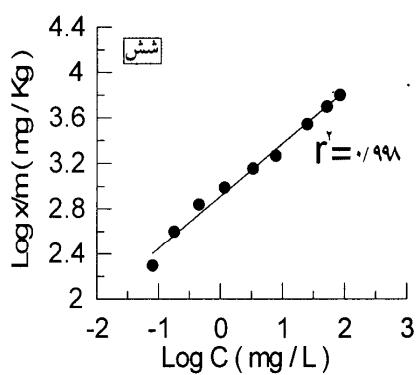
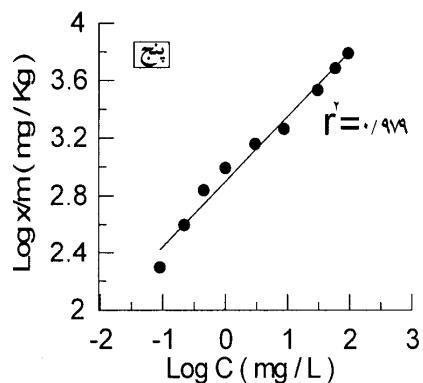
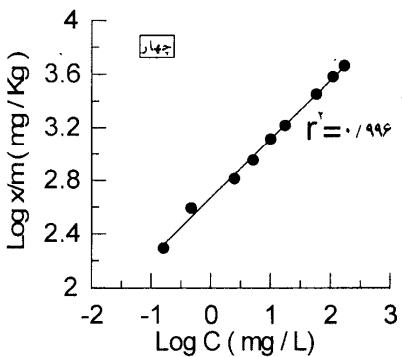
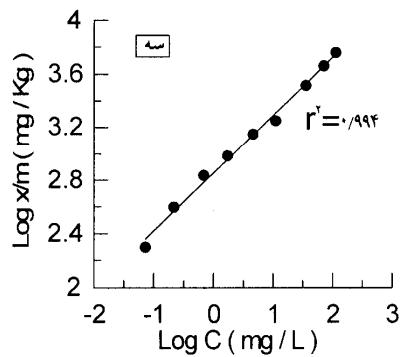
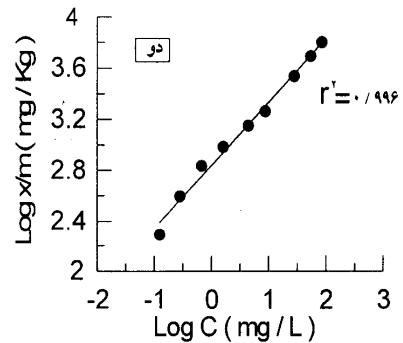
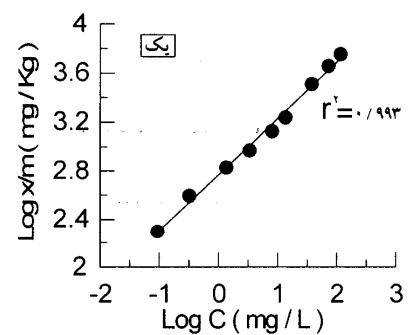
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( $p < /$ ) ( $R$ ) ) ( $pH = /$ ) ( $pH = /$ ) ( $pH = /$ )

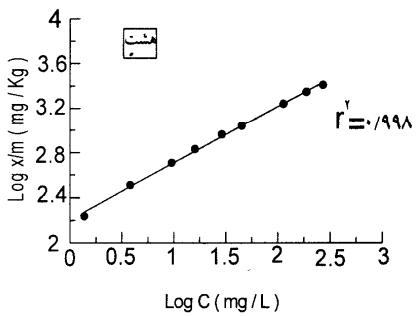
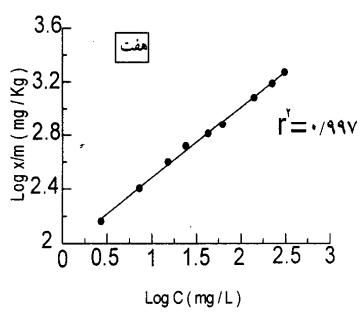
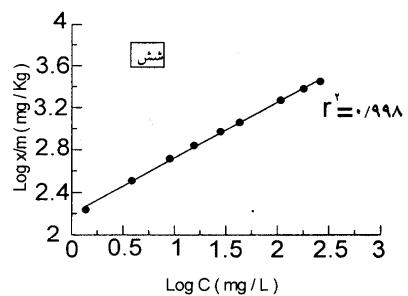
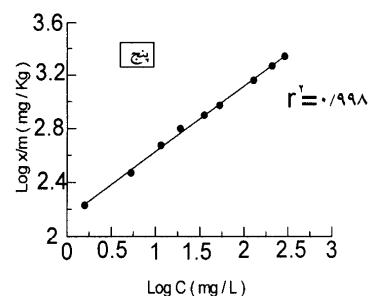
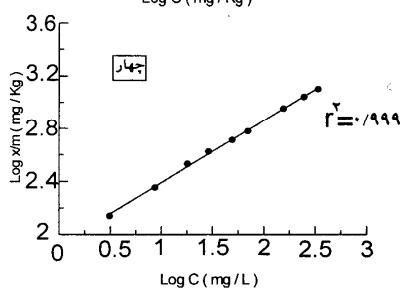
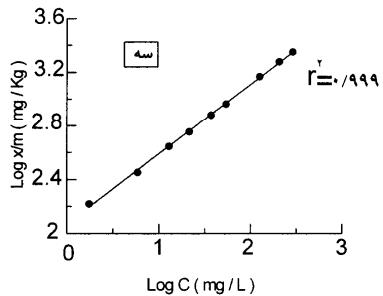
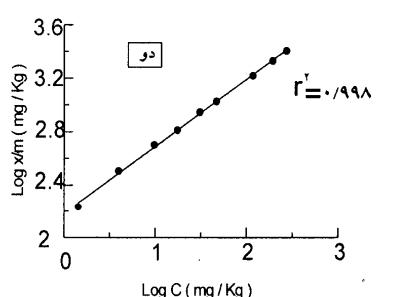
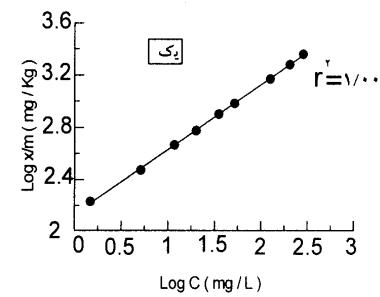
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(	)	(	)	(	)	(+)	)	(-)	pH	EC dSm <sup>-1</sup>
/	/	/					/	/	/	Torrifluvents
/	/	/		/			/	/	/	Haplocalcids
/	/	/			/	/		/	/	Haplocalcids
/	/	/		/			/	/	/	Haplocalcids
/	/	/		/	/	/		/	/	Haploustepts
/	/	/				/		/	/	Haploustepts
/	/	/				/		/	/	Haploustepts
/	/	/				/		/	/	Haploustepts

/ \*\* / / \*\* / /







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( )

$$\begin{array}{c}
K_f \quad (K_f) \\
\left( \frac{1}{n} \right) \\
K_f \quad / \quad / \quad \left( \frac{1}{n} \right) \\
/ \quad / \quad ( ) \quad / \\
K_{f=} \quad / + / \text{CEC} + / \% \text{Clay} - \% \text{OM} \\
R = / \quad ** \\
K_f \quad / \quad / \quad \left( \frac{1}{n} \right) \\
( ) \quad K_f \quad ( ) \\
/ \quad / \quad / \\
K_f \quad / \quad / \quad / \\
/ \quad / \quad / \\
K_{f=} \quad / + / \text{CEC} \quad R = / \quad ** \\
K_{f=} \quad / + / \% \text{Clay} \quad R = / \quad ** \\
K_{f=} \quad / + \% \text{OM} \quad R = / \quad * \\
K_{f=} \quad - / \% \text{CaCO}_3 \quad R = / \quad *
\end{array}$$

$$\begin{array}{c}
( ) \quad ( ) \\
K_f \quad (K_f) \\
( CEC )
\end{array}$$

$$\begin{array}{c}
K_{f=} \quad + / \text{CEC} \quad R = / \quad * \\
K_{f=} \quad + / \% \text{Clay} \quad R = / \quad *
\end{array}$$

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