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.(Holik, 2006)
                                                       (Nie et al.,
                 (ONG
                             ONP)
                                                                                                  .1997)
                                                                                    (
                Renders (1995)
Ackermann
                                                       Costa
                                          (2000)
                                                                                        .& Rubio (2005)
 H_2O_2 + H_2O \longrightarrow HOO^- + H3O^+
                                                                                                       )
 H_2O_2 + OH^- \longrightarrow H_2O + HO_2^-
                                                                  .(Mirshokraei, 2001)
                                                                                                       )
                                                             (DTPA)
                .(Helming et al., 1985)
                                                             .(Mc Cormick, 1990)
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.(Ackermann, 2000) Rahmaninia et al., (2004) NaOH .(Bajpai, 2005) Ruzinsky & Bennington (2005) . Van lierop (1993) .(McCormick, 1990) COD Mirshokraei,) .(2001 H_2O_2 NaOH Bajpai)) NaOH .(2005 H_2O_2 DTPA

¹ Chemical Oxygen Demand

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/ / ) NaOH
                               NaOH H<sub>2</sub>O<sub>2</sub>
                                                    / / / / ) H<sub>2</sub>O<sub>2</sub> (
                                                                  (MOW) (
                                                             (OMG)
                          TAPPI T205 om-88
                                                                                      (ONP)
TAPPI T452
Elrepho 2000
                                      om-98
ISO5350-
          1
                                        2:06
                                                                     % /
                                                                                           % / )
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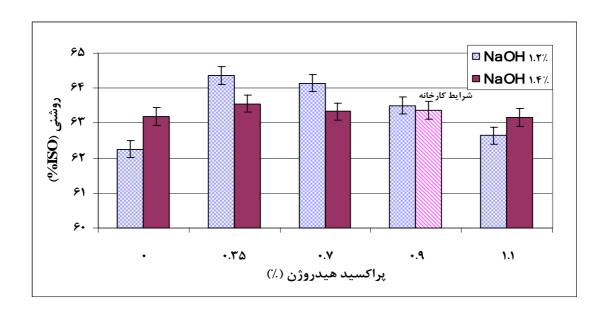
P	P	P	
ns	**	**	
/ ns	/ ns	ns	
/ ns	**	**	×

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۶۴.۵ a شرایط کارخانه a T 94 روشنی (OSI%) ۶۳.۵ b ۶٣ ۶۲.۵ 82 ۶۱.۵ ۰.۳۵ ٠.٧ ٠.٩ 1.1 پراکسید هیدروژن (٪)

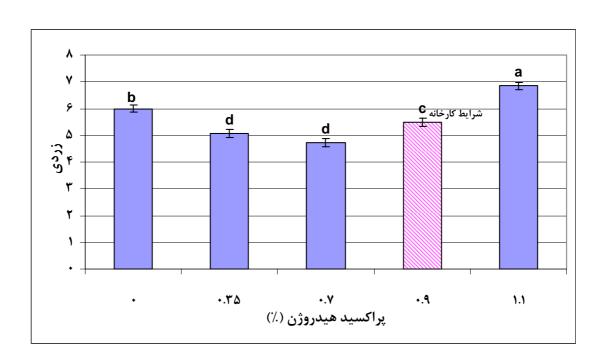
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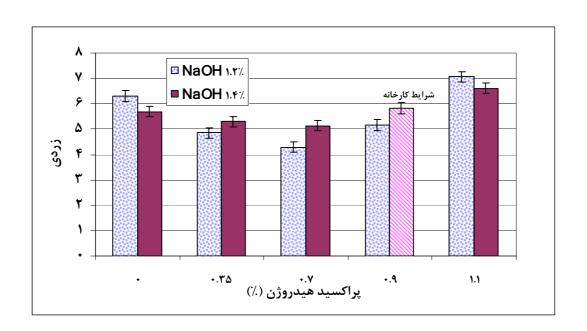


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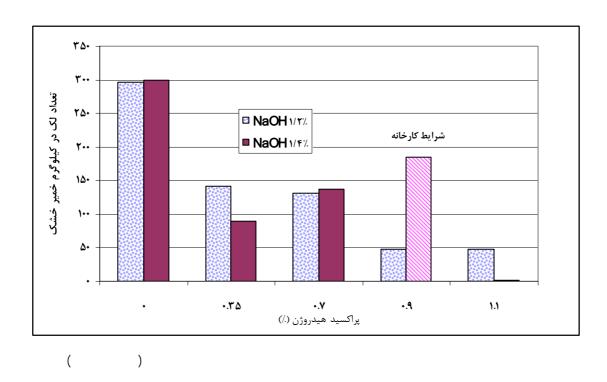


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$$Y = / A + (/ B) + (/ C) ($$

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(%) (%)

.(Garver, 2000) . (HOO-) % / .(Hipolit, 1992)

Ferguson (1992)
HOO-

рН

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Investigation on the Optimal Dosage of Hydrogen Peroxide and Sodium Hydroxide to Improve Flotation Deinking Efficiency for Mixed Waste Papers

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Abstract

Flotation deinking is the prominent technology for ink removal from waste paper. In this study, flotation deinking was carried out applying two levels of sodium hydroxide and five levels of hydrogen peroxide on a mixture of waste papers consisting of: 30, 30 and 40 weight percent OMG, ONP and MOW, respectively. The deinking efficiency was evaluated based on brightness, yellowness and dirt count of hand sheets. The results showed that increasing hydrogen peroxide dosage can increase hand sheets brightness, however, at higher hydrogen peroxide levels, the brightness was reduced probably due to the presence of more NaOH and the limitation of hydro peroxide efficiency for chromophore elimination. Optimal ratio of H₂O₂/NaOH, significantly increase the efficiency of deinking at lower alkali usage. This research showed that higher deinking efficiency may not be necessarily reached applying more chemicals, but optimal chemicals ratio plays an important role in deinking efficiency.

Keywords: Flotation deinking, Waste paper recycling, Hydrogen peroxide, Sodium hydroxide, Brightness, Yellowness, Dirt count

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